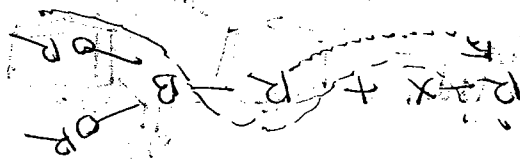
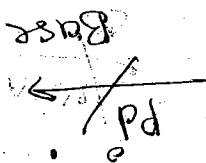
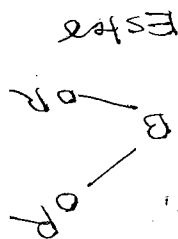
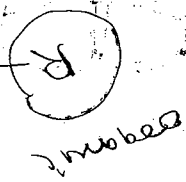


$$\{R - R'\}$$


• proof



organic Boranes → Acids

→ Barium sses amns II

coldest morning for a good long time in season

- In org. synthesis,  $\text{CH}_3\text{COOH}$  is used in 0 and 2.  $\text{CH}_3\text{COOH}$  is used in 0 and 2.

to be good  
in number

पुनर्विचार = 27

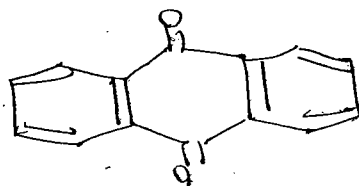
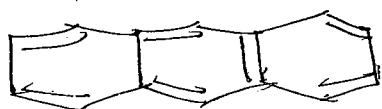
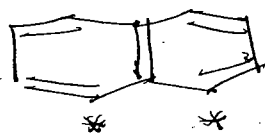
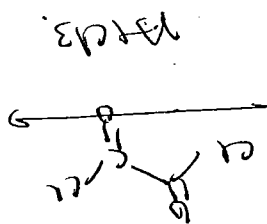
•  $\rho dV$

11/10/17

$$\text{Pd}(\text{PPh}_3)_4, \quad \text{Pd}(\text{Core})_2, \quad \text{Pd}(\text{PPh}_3)_2$$

\* Pd catalysed reaction:-  $\text{Pd}^0 / \text{Pd}$

## Coupling Reactions :-



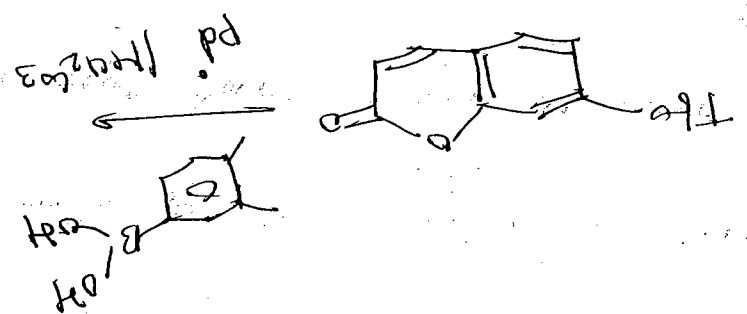
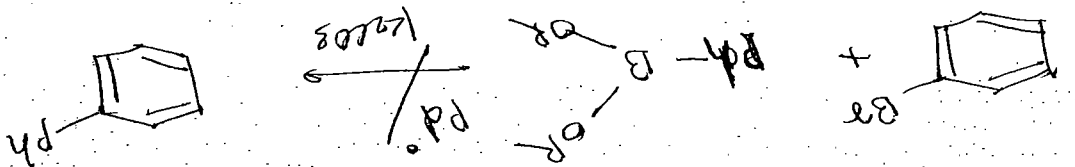
$R' = \text{aryl, alkyl, alkynyl}$

$R = \text{acetyl, allyl, alkyl, alkynyl}$

$X = \text{H, OH, OTS, OTf, etc.}$

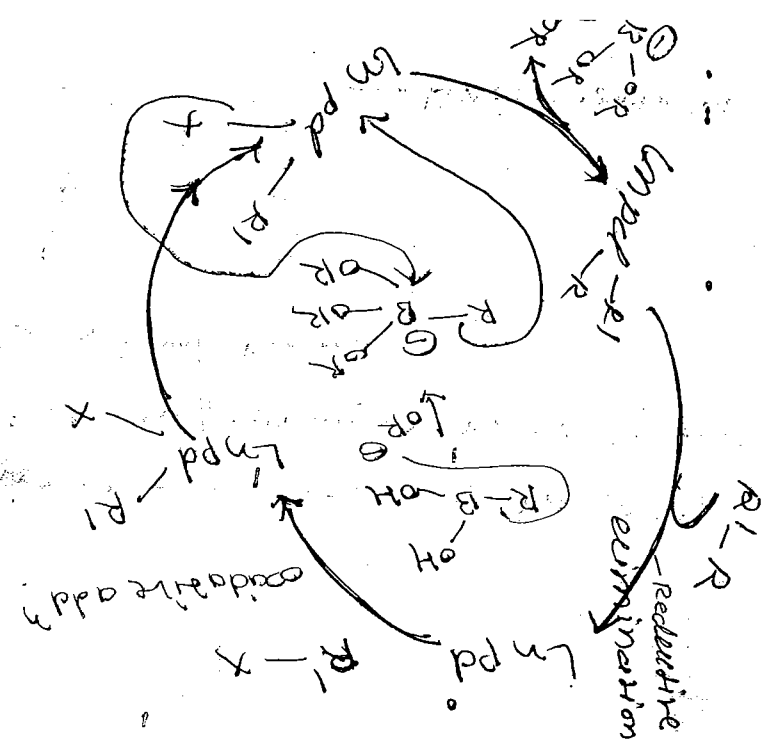
$Cl, Br, I$

Base =  $Na_2CO_3, K_2CO_3, Cs_2CO_3, Na^+ Na^+ \text{ etc.}$



useful points:-

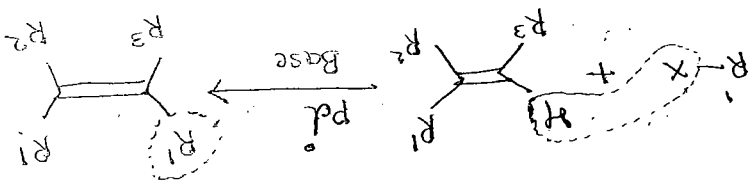
Mechanism:-



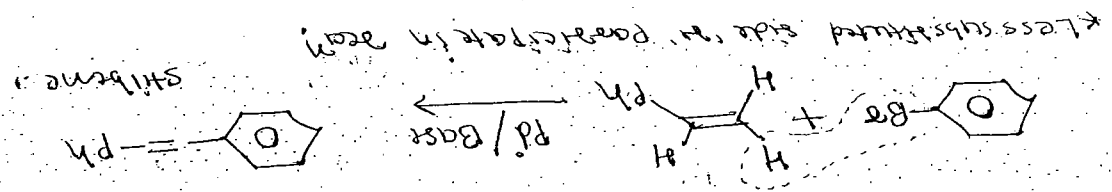
- \* High % of yields
- \* Organo borane (base)
- \* Acetate anion
- \* Not sensitive to moisture
- \* It has a triaryl containing
- \* Other functional
- \* group unaffected.

- Easy to remove by product.

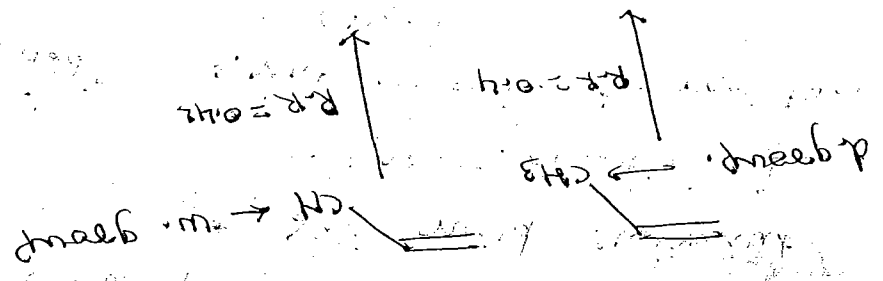
\* Heck-cross coupling:-



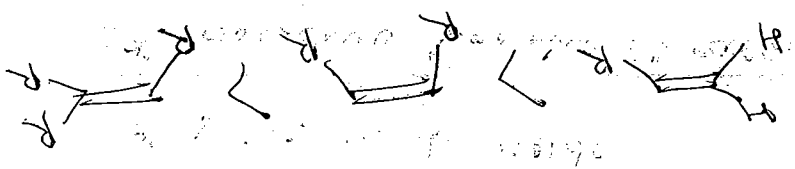
R' = Any org. group i.e. Aryl, Alkenyl, alkyl.  
 X = Halo group, OTs, OTf, H<sub>2</sub><sup>+</sup>  
 Base = NaOAc, NaOH, K<sub>2</sub>CO<sub>3</sub> (mild)



- Pd catalyzed coupling between organic halo compounds and triaryl boronic acids.
- Pd catalyst to carry out. High % of yields.
- Poor by products easy to remove. more sensitive at a low and moisture.
- Tolerates many other organic functions. ester and carbonyl etc.
- Alkene is unsymmetrical then places place at less substituted position.



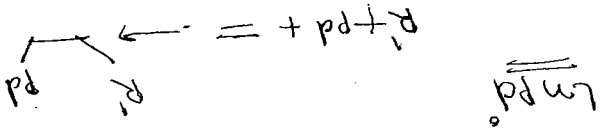
Rate of atom unaffected by donating or withdrawing groups on alkenes but is same as the withdrawing group increases rate of atom. compare to donating group.



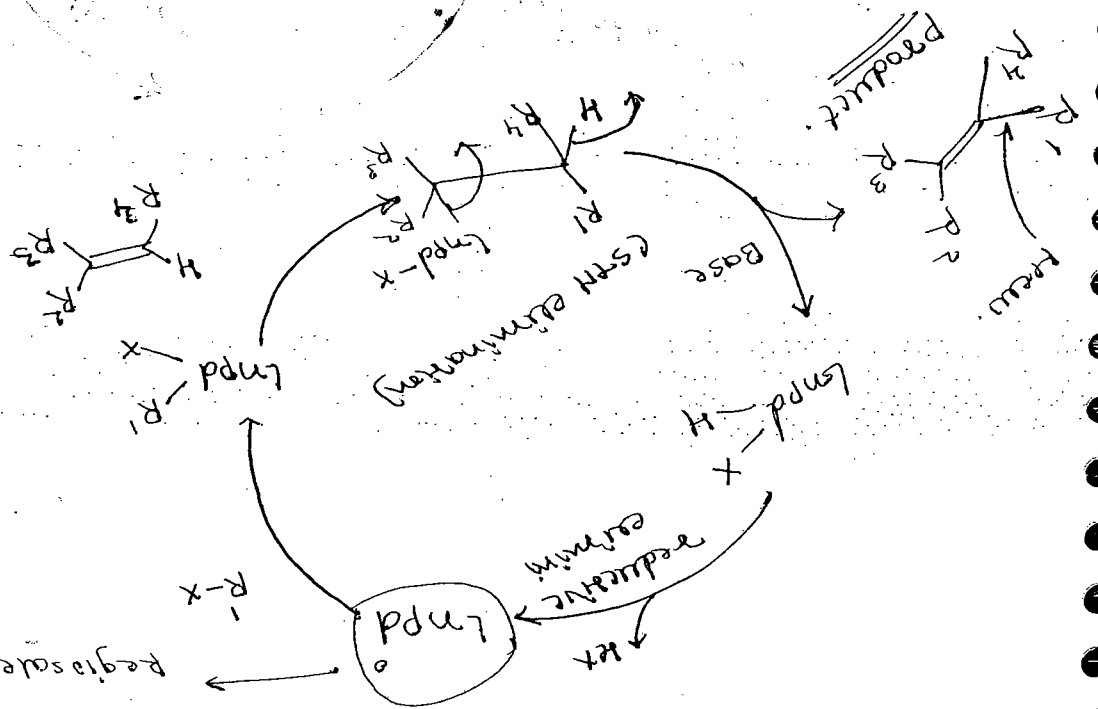
- In increasing no. of substituents on alkenes, rate of coupling decreases.

Pathway:-

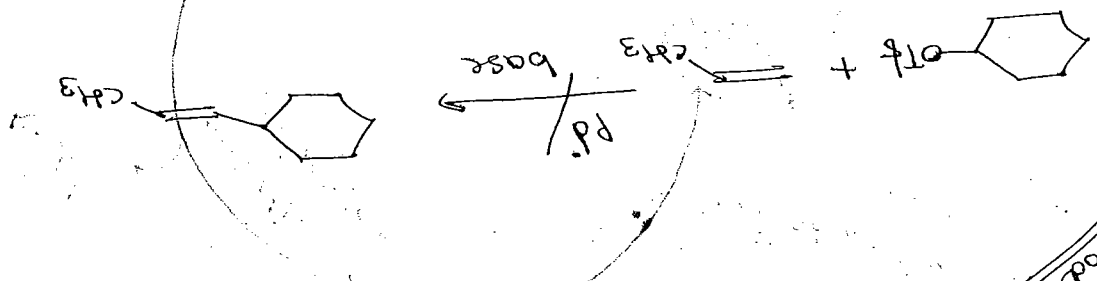
- It is in a cyclic manner.



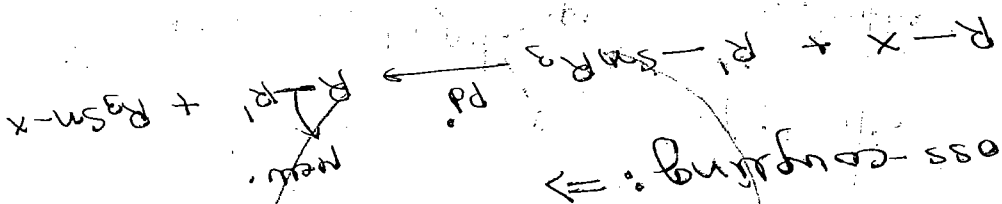
Regioselective oxidation.



e.g.:



\* STILL cross-coupling:



- Pd catalysed coupling between organic halo comp. with organo

Sn compounds still cross coupling.

- It's similar to Suzuki-cross coupling.

- Easy to carry out. Yields of the rxn. not

sensitive to the nucleophile and etc.

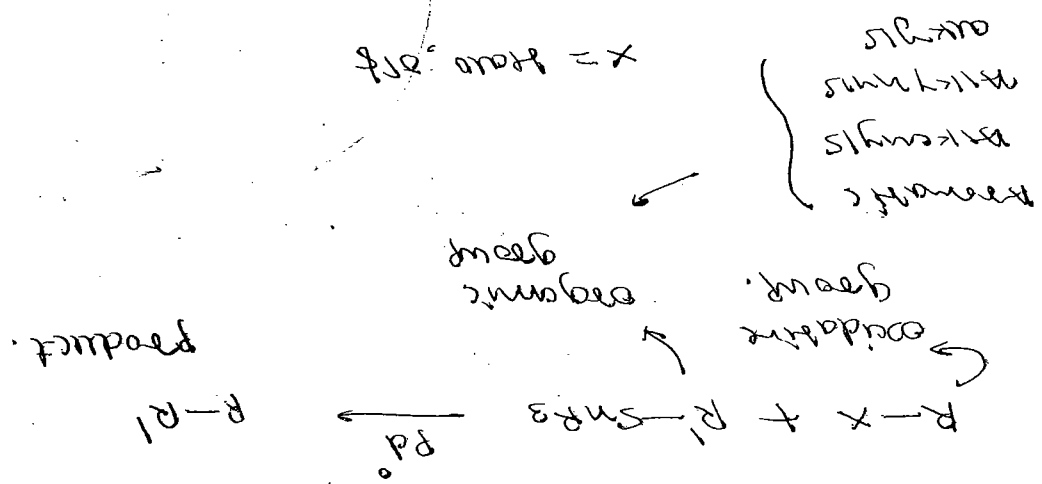
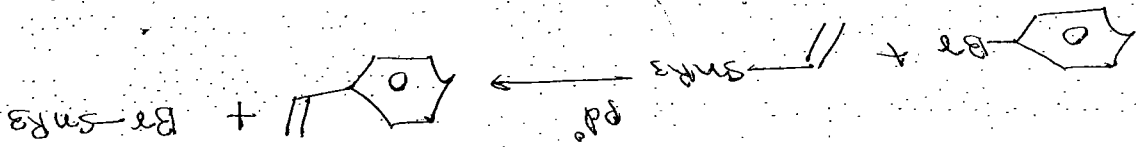
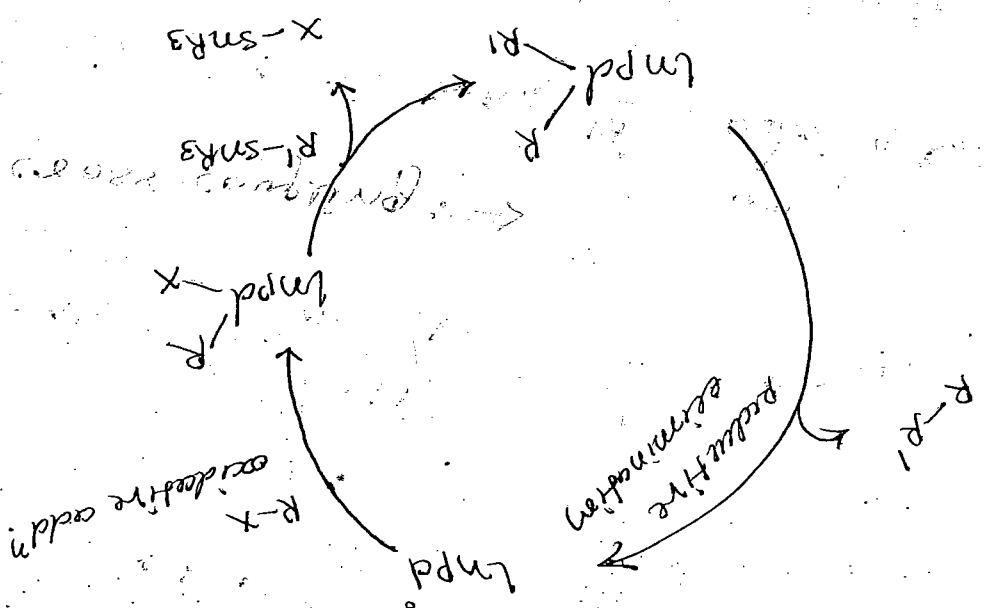
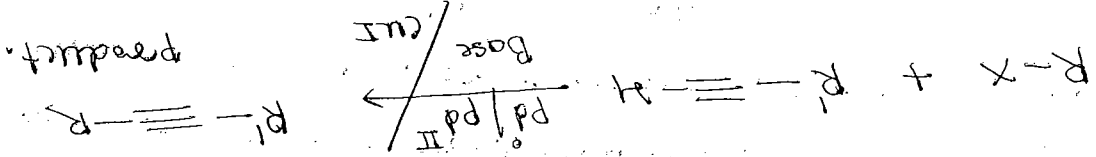
- tolerates other organic functions i.e. acid, ester, amide, ketone etc.

\* - Sn compound expensive. Environment pollutant.

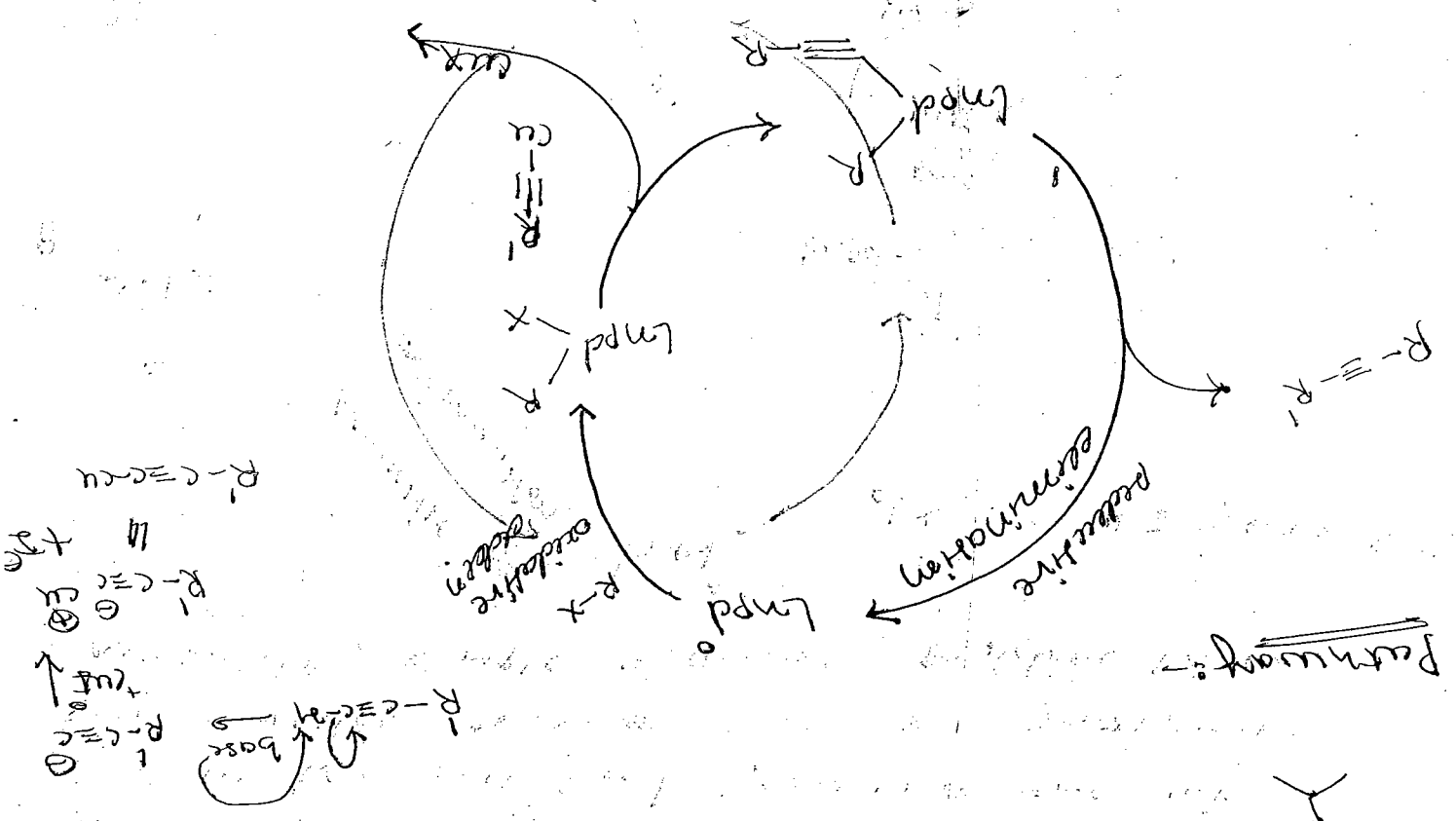
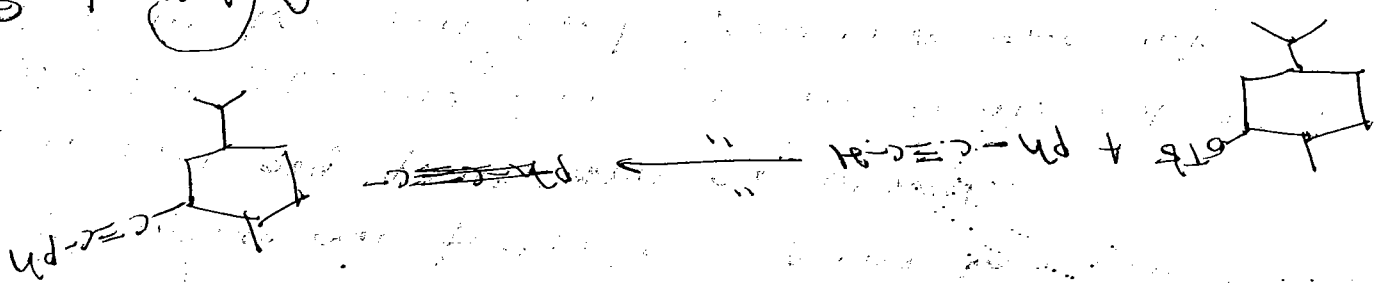
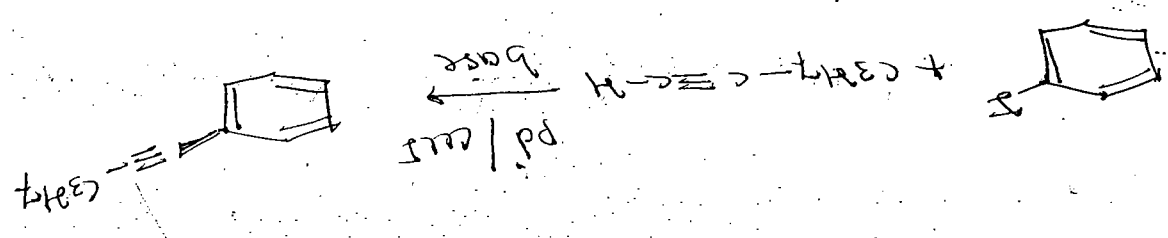
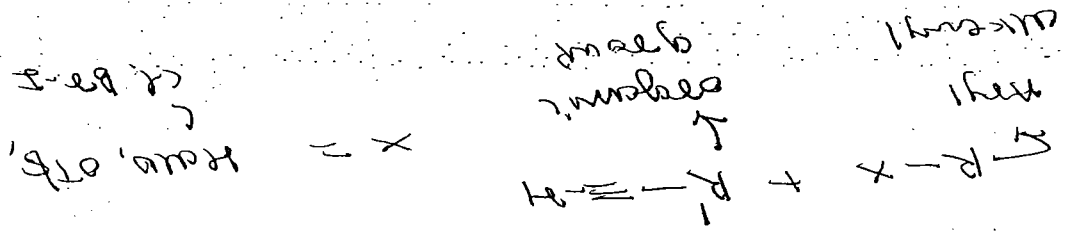
Pathway on next page =>

- Coupling betn or Pd-Cu catalysed coupling betn
- Organos have comp. or triplots with terminal groups in basic medium Sonogashira coupling
- Terminal alkyn to internal alkyn
- Terminal alkyn can convert to Cu salt. Copper salt

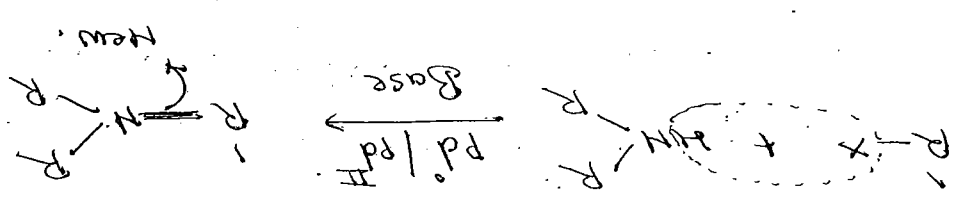
**Sonogashira coupling :-**



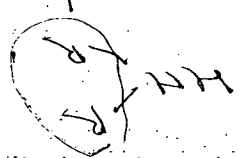
- copper salt undergoes Pd-complex.
- Not sensitive air and the
- tolerate other functional group. good % of yield.
- product easy to isolate.



\* BUCHHARDT - HERTWIG COUPLING : C-N bonding, @



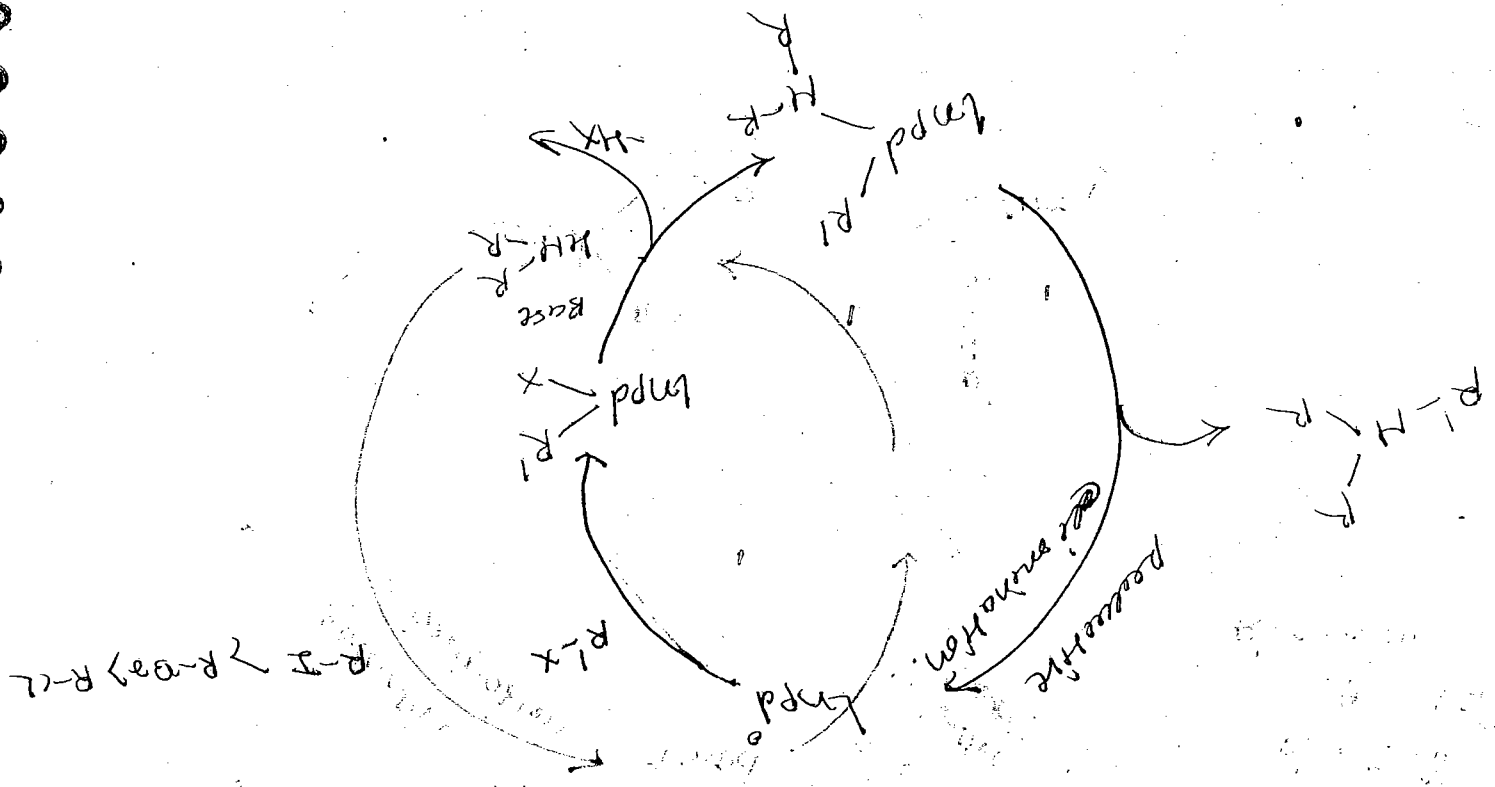
- Pd catalysed coupling between organic halo comp. or triazoles and 1° and 2° amines in basic medium called Buchwald HERTWIG coupling.



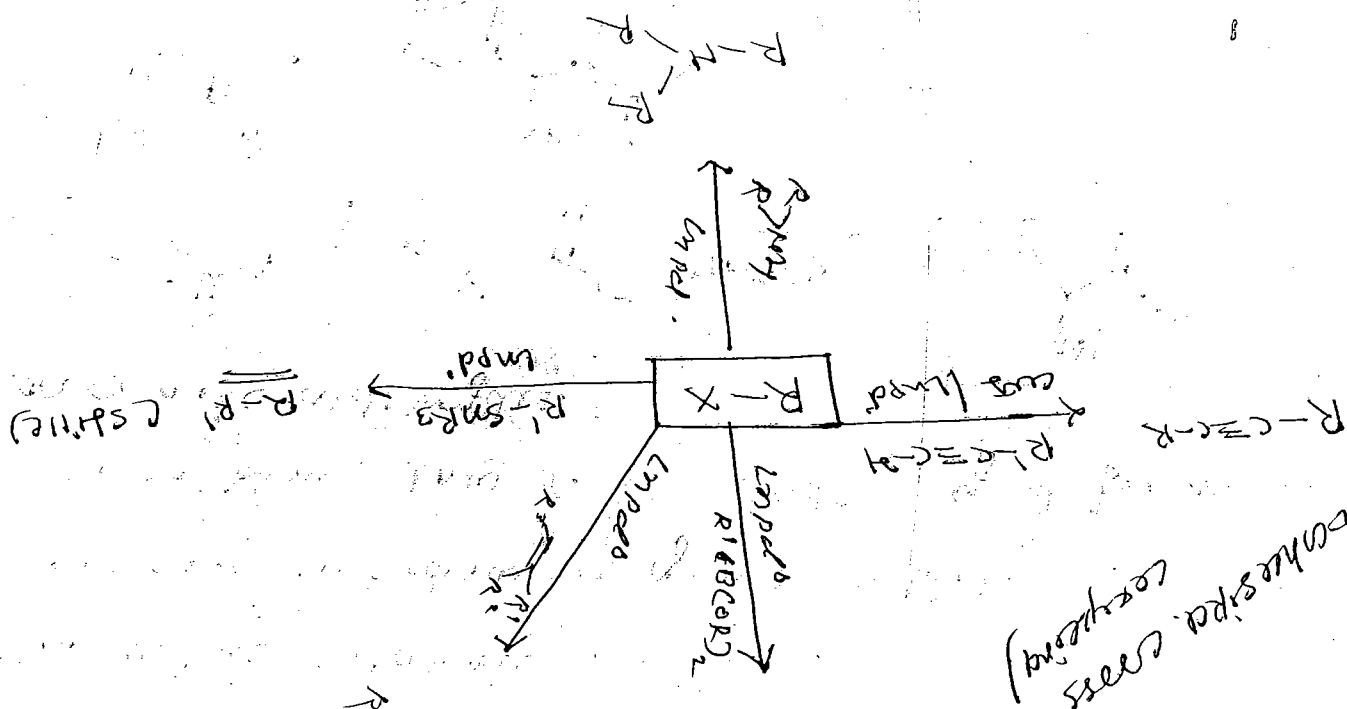
any/alkenyl  
any organic group

X = Cl, Br, I (Haw)  
etc.

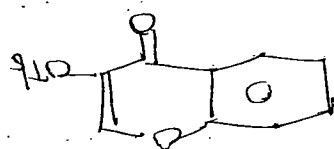
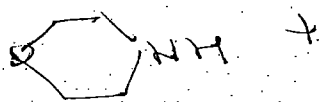
- 2° amines are faster in coupling in comparison to 1° amine.
- Reduced org. group, aryl, or alkenyl.
- Good % of yield, not sensitive to air and also moisture.
- C-N bond formation even with heterocycles / common bases are hydroxides, alkoxides, simple 3° amines pyridine triamine.



(Quaternary)



(Hetero cross)



mercaptide

