

Dayanand Science College, Latur
Department of Physics
Model Question Paper (MCQ for Practice)

Paper Name: Atomic, molecular and nuclear physics

Paper No.: XIV

SEM-VI

1. The magnitude of spin quantum number is always....
 - A. $1/2$
 - B. -1
 - C. 1
 - D. None of these

2. m_j can have only values from $-j$ to j , excluding zero.
 - A. $j(j+1)$
 - B. $2j+1$**
 - C. Both (A) & (B)
 - D. None of these

3. The maximum number of electron in a shell is
 - A. N
 - B. $2n^2$**
 - C. $3n^2$
 - D. $2n$

4. In the spectra of alkali metals, the doublet separation decreases with
 - A. Increase in atomic number
 - B. Decrease in orbital number
 - C. Increase in principal quantum number**
 - D. Dalton 's atomic theory

5. When the atom is placed in external weak magnetic field, then we observe.....
- A. Stark effect
 - B. Anomalous Zeeman effect**
 - C. Normal Zeeman effect
 - D. None of these
6. The effect of external magnetic field on spectral lines is known as
- A. Zeeman effect
 - B. Stark effect**
 - C. Paschen-Back effect
 - D. Raman effect
7. The spin quantum number refers to.....
- A. Energy of electron**
 - B. Direction of electron spin
 - C. Sublevel or shape of the orbital
 - D. Orientation of orbital nucleus
8. In strong field star effect for $n = 2$, the degeneracy in fine structure is lifted to levels.
- A. 1
 - B. 2
 - C. 3
 - D. 4**
9. An electron in 'f' subshell can have a principle quantum number...
- A. 0
 - B. 2**
 - C. 1
 - D. 3
10. The selection rule for the normal Zeeman effect is....

- A. $\Delta M_j = 0, \pm 1$
- B. $\Delta M_l = 0, \pm 1$**
- C. $\Delta S = 0, \Delta L = 0, \pm 1$
- D. $\Delta S = 0, \Delta L = 0, \Delta J = 0, \pm 1$
11. The molecular system can be stable if the total energy possessed by the molecule is.....
- A. Zero
- B. Minimum**
- C. Maximum
- D. None of these
12. The rotational kinetic energy of a diatomic molecule for $j = 0$ is
- A. $2 B$
- B. $6 B$
- C. $0 B$**
- D. $12 B$
13. A pure rotational spectrum occurs in..... region.
- A. Ultraviolet
- B. Infra-red
- C. x-ray
- D. Microwave**
14. An electronic spectrum of diatomic molecules occurs in region.
- A. Uv-vis**
- B. Infra-red
- C. x-ray
- D. Microwave
15. Raman shift occurs in region.
- A. Microwave
- B. Infra-red**

- C. UV region
D. X-ray region
16. To observe Raman effect, molecules must be
- A. Polar
B. Non-polar
C. Both (A) and (B)
D. None of these
17. The intensity of Rayleigh's line is..... as compared to Raman lines.
- A. Zero
B. Very low
C. **Very high**
D. None of these
18. The zero point energy of a molecule is
- A. $\frac{1}{2} h\nu_0$
B. **$h\nu_0$**
C. Zero
D. None of these
19. Which type of scattering result in a longer wavelength than the incident light?
- A. Anti-Stokes
B. **Stokes**
C. Rayleigh
D. All of the above
20. Which of the following has the maximum penetrating power?
- A. Radio waves

- B. Microwaves
 - C. Infrared rays
 - D. Gamma rays**
21. Nuclear fission is the phenomenon of
- A. Light nucleus splitting
 - B. Heavy nucleus combining
 - C. Heavy nucleus splitting**
 - D. Light nucleus combining
22. The average energy of a neutron produced in fission of Uranium 235 isotope is
- A. 3MeV
 - B. 1MeV**
 - C. 2MeV**
 - D. 10MeV
23. Which isotope of Uranium has the capacity to sustain the chain reaction?
- A. U-235**
 - B. U-230
 - C. U-231
 - D. U-245
24. The energy we get in nuclear reaction comes from.....
- A. The mass of the fuel**
 - B. The sun
 - C. Water
 - D. None of these
25. The Q-value of fission reaction is of the order
- A. 200MeV**
 - B. 1MeV

- C. 300MeV
- D. None of these
26. Name the moderator used in the nuclear reactor?
- A. Plutonium
- B. Thorium**
- C. Graphite
- D. Berilium
27. A nucleus of medium mass with excess of neutrons may decay with the emission of...
- A. Positron
- B. Electron
- C. Proton
- D. Neutron**
28. A typical beta (β) decay chain is.....
- A. ${}^{140}_{54}\text{Xe} \rightarrow {}^{140}_{55}\text{Cs} \rightarrow {}^{140}_{56}\text{Ba} \rightarrow {}^{140}_{57}\text{La} \rightarrow {}^{140}_{58}\text{Ce (Stable)}$
- B. ${}^{140}_{54}\text{Xe} \rightarrow {}^{140}_{55}\text{Cs} \rightarrow {}^{140}_{56}\text{Ba} \rightarrow {}^{140}_{57}\text{La} \rightarrow {}^{140}_{58}\text{Ce (Unstable)}$
- C. ${}^{140}_{54}\text{Xe} \rightarrow {}^{141}_{54}\text{Cs} \rightarrow {}^{142}_{54}\text{Ba} \rightarrow {}^{143}_{54}\text{La} \rightarrow {}^{144}_{54}\text{Ce (Stable)}$
- D. ${}^{140}_{54}\text{Xe} \rightarrow {}^{141}_{54}\text{Cs} \rightarrow {}^{142}_{54}\text{Ba} \rightarrow {}^{143}_{54}\text{La} \rightarrow {}^{144}_{54}\text{Ce (Unstable)}$**
29. The safety-rods present to shut down the reactor are made up of.....
- A. Cadmium**
- B. Calcium
- C. Carbon
- D. **None of these**
30. Which of the following is not used as a moderator in a nuclear reactor?
- A. H₂O
- B. D₂O
- C. C
- D. Al**

31. The nuclear reaction $4\text{}^1_1\text{H} \rightarrow \text{}^4_2\text{He} + 2\text{}^0_{-1}\text{e} + 26\text{MeV}$ represent;
- γ -decay
 - Fission
 - Fusion**
 - None of these
32. Nuclear fusion required high temperature because.....
- All nuclear reactions absorb heat
 - The mass deficit must be supplied
 - The binding energy must be supplied from an external source
 - The particles cannot come closer unless they are moving rapidly**
33. The method of carbon dating works because.....
- C^{14} has higher atomic weight than C^{12}
 - C^{14} is a stable isotope**
 - C^{14} content of the dead body increases with time because of cosmic ray bombardment
 - None of the above
34. The reaction $e^+ + p^+ \rightarrow \nu_e + \pi^+$ forbidden because of.....
- Law of electron number conservation
 - Law of baryon number conservation**
 - Law of momentum energy conservation
 - Law of muon number conservation
35. The quarks are supposed to exist in following number of flavours...
- Two
 - Four
 - Six
 - Eight
36. Suppose that a neutron at rest in free space decays into a proton and electron. This

process would violate....

- A. Conservation of charge
 - B. Conservation of energy
 - C. Conservation of linear momentum**
 - D. Conservation of angular momentum
37. Nuclear energy can be made available;
- I. By the fission of certain heavy nuclei
 - II. By the fusion of very light nuclei
- A. Only (I)
 - B. Only (II)
 - C. Both (I) and (II)**
 - D. None of these
38. Which of the following is correct:
- A. $\Delta S = 0$ for the reaction;
 $\Sigma^+ \rightarrow \Lambda^0 + e^+ + \nu_e$
 - B. Strangeness for Σ is -2
 - C. Strangeness zero for nucleons and non-zero for hyperon
 - D. All of the above**
39. Isospin numbers are associated with.....
- A. Leptons
 - B. Mesons
 - C. Hadron**
 - D. All of the above
40. Which of the following decay is forbidden?
- A. $\mu^- \rightarrow e^- + \nu_\mu + \bar{\nu}_e$
 - B. $\mu^- \rightarrow e^+ + e^- + e^-$**
 - C. $\pi^+ \rightarrow e^- + \nu_e$
 - D. $\pi^+ \rightarrow \mu^+ + \nu_\mu$